

阿伏伽德罗常数的 改进与新式测量

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Amedeo Avogadro
(1776-1856)



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(1776-1856)

$N_A =$



$\dots 2140857(74) \times 10^{23} \text{ mol}^{-1}$

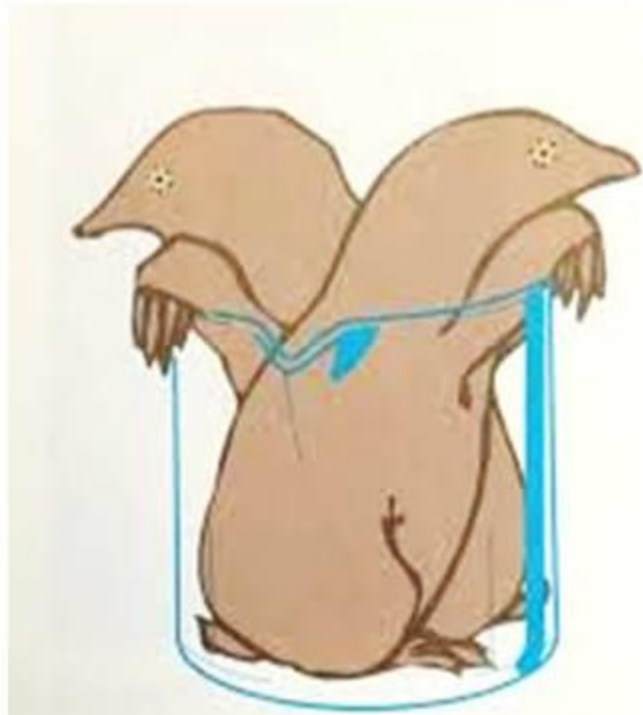


Figure 8
Two moles per litre





John Dalton
(1766-1844)



Joseph Louis Gay-Lussac
(1778-1850)



John Dalton
(1766-1844)



Joseph Louis Gay-Lussac
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“Constant”

- Avogadro Hypothesis



John Dalton
(1766-1844)



Joseph Louis Gay-Lussac
(1778-1850)



Joseph Loschmidt
(1721-1895)

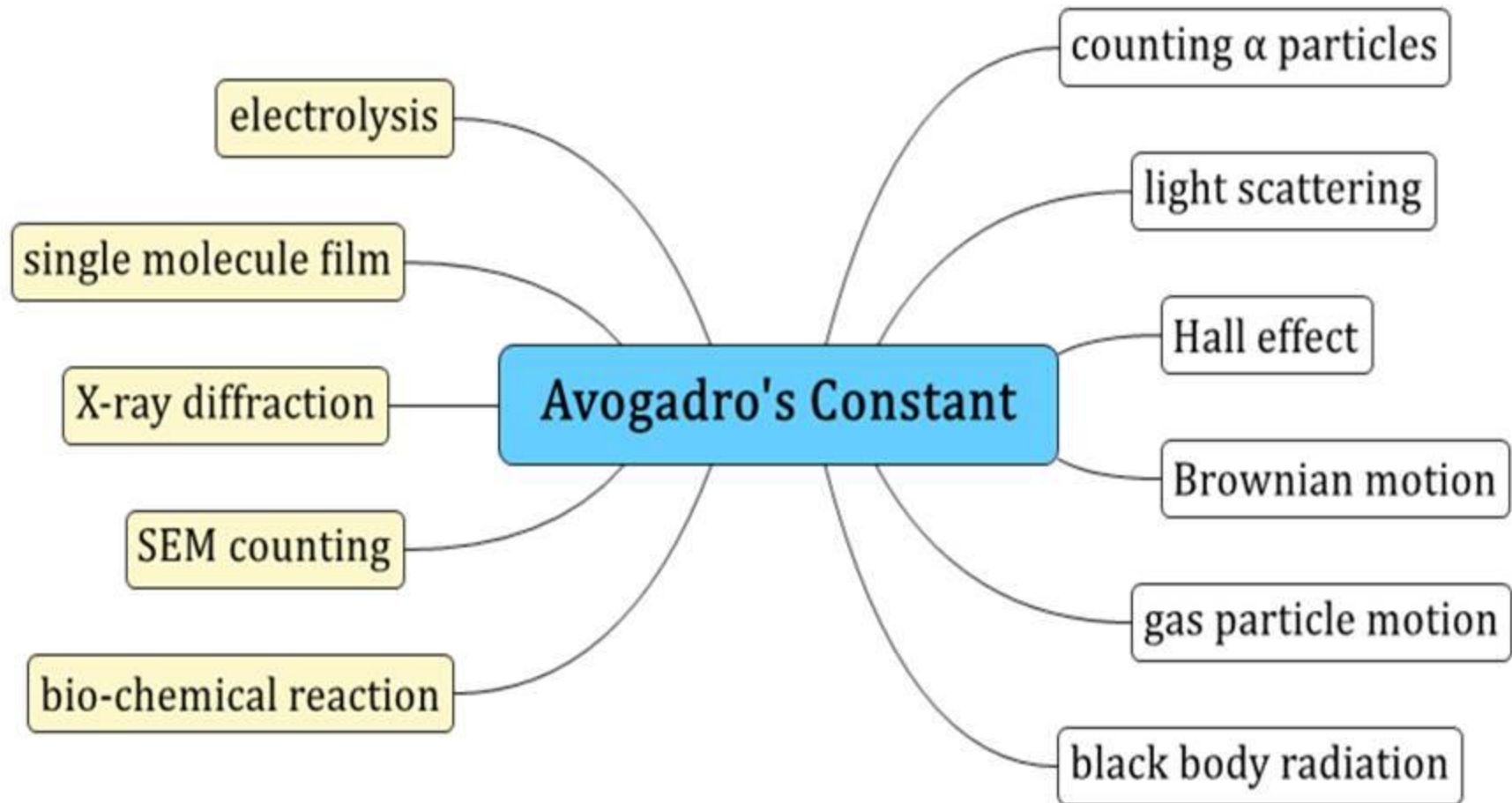
“Constant”

- Avogadro Hypothesis

$$1.0 \times 10^{24} \text{ mol}^{-1}$$

Table 1 Partial values reported for N_A over the history

Year	Author	$N_A/10^{23}$
1908	Perrin	6.7
1917	Millikan	6.064
1929	Birge	6.0644
1945	Birge	6.02338
1951	DuMond	6.02544
1965	Bearden	6.022088
1973	Cohen	6.022045
1987	Deslattes	6.022134
1994	Basile	6.0221379
2001	De Bièvre	6.0221339



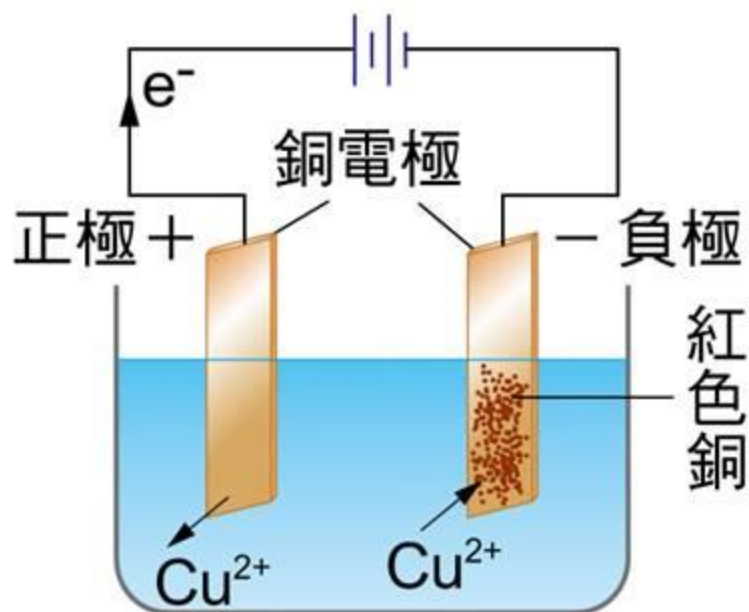
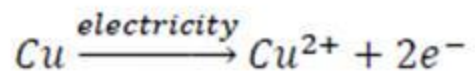
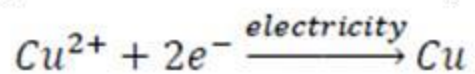


Figure 1 Illustration for electrolysis



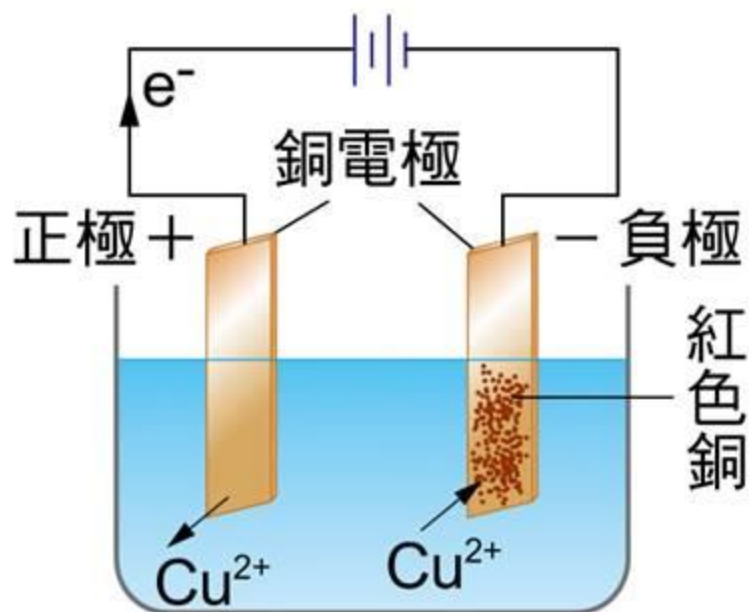
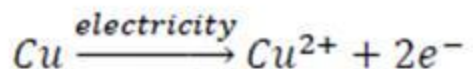
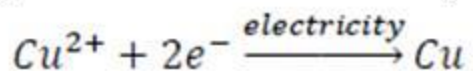


Figure 1 Illustration for electrolysis



$$\begin{cases} F = N_A e \\ m = \frac{QM}{zF} \end{cases} \Rightarrow N_A = \frac{QM}{2me}$$

$$\begin{aligned} * M &= 63.55 \text{ g} \cdot \text{mol}^{-1} \\ e &= 1.60218 \times 10^{-19} \text{ C} \end{aligned}$$

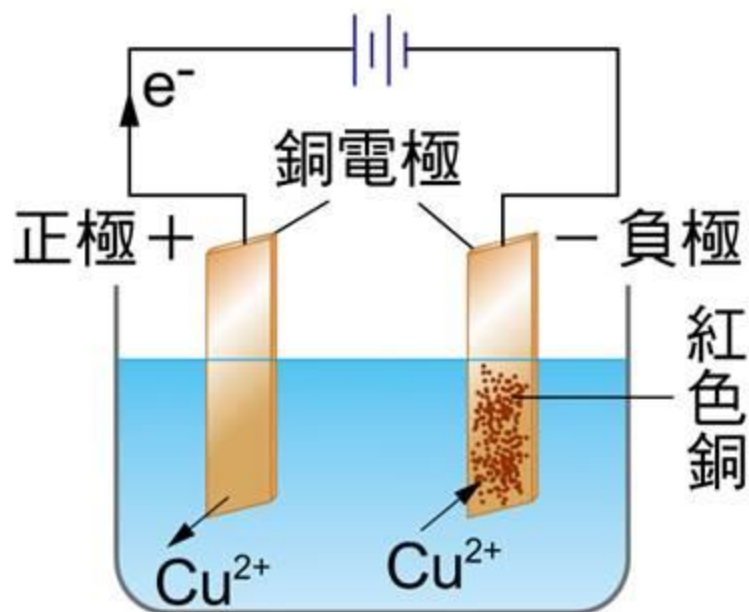
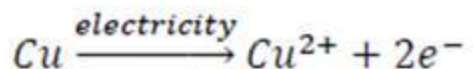
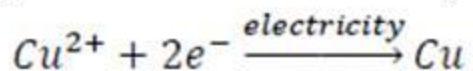


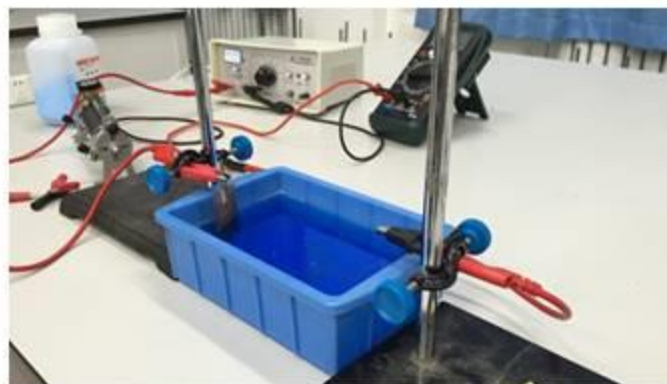
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$$Q = \sum I \cdot \Delta t$$



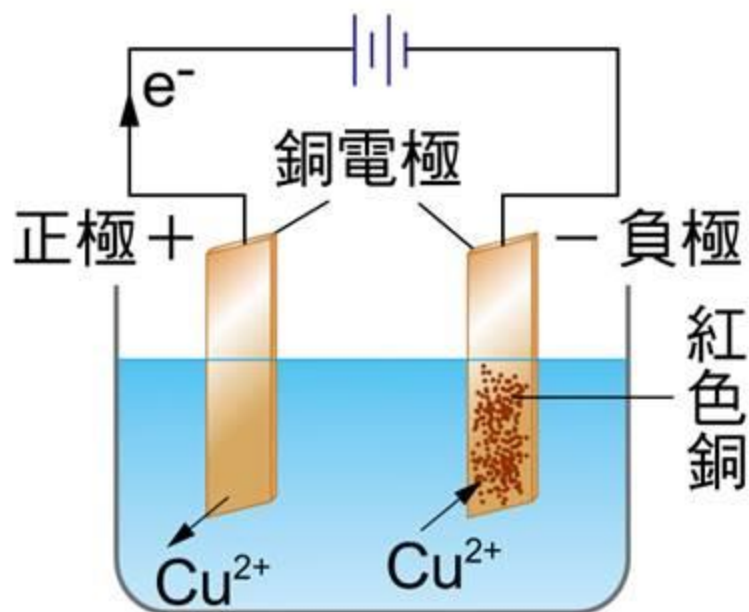
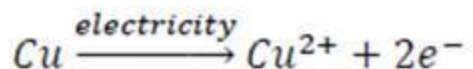
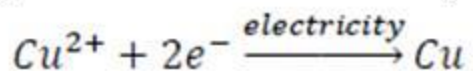


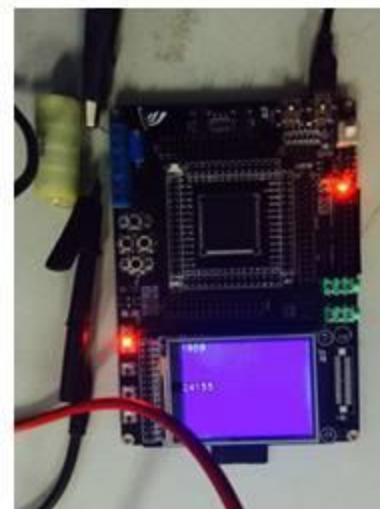
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$$Q = \int I \cdot dt$$



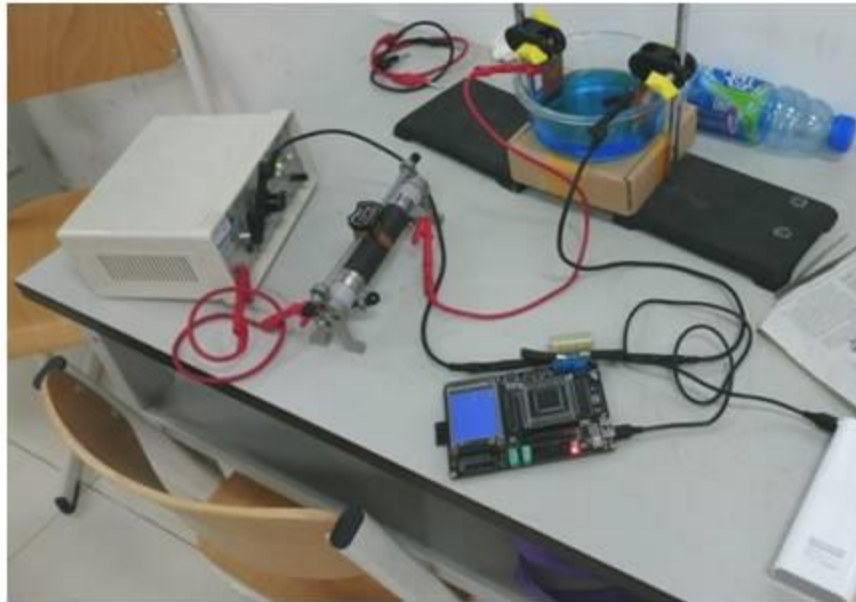
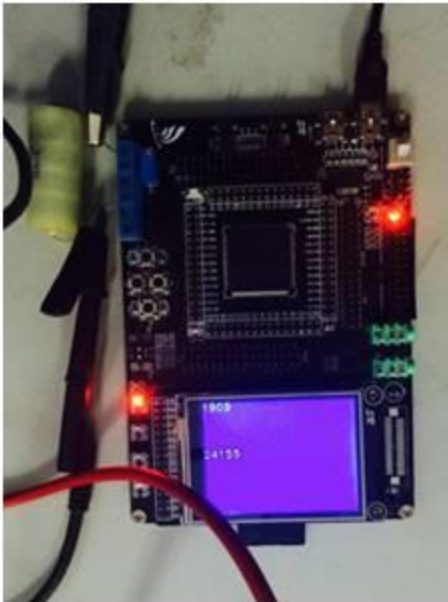


Figure 2 Coulomb meter

/0.1s

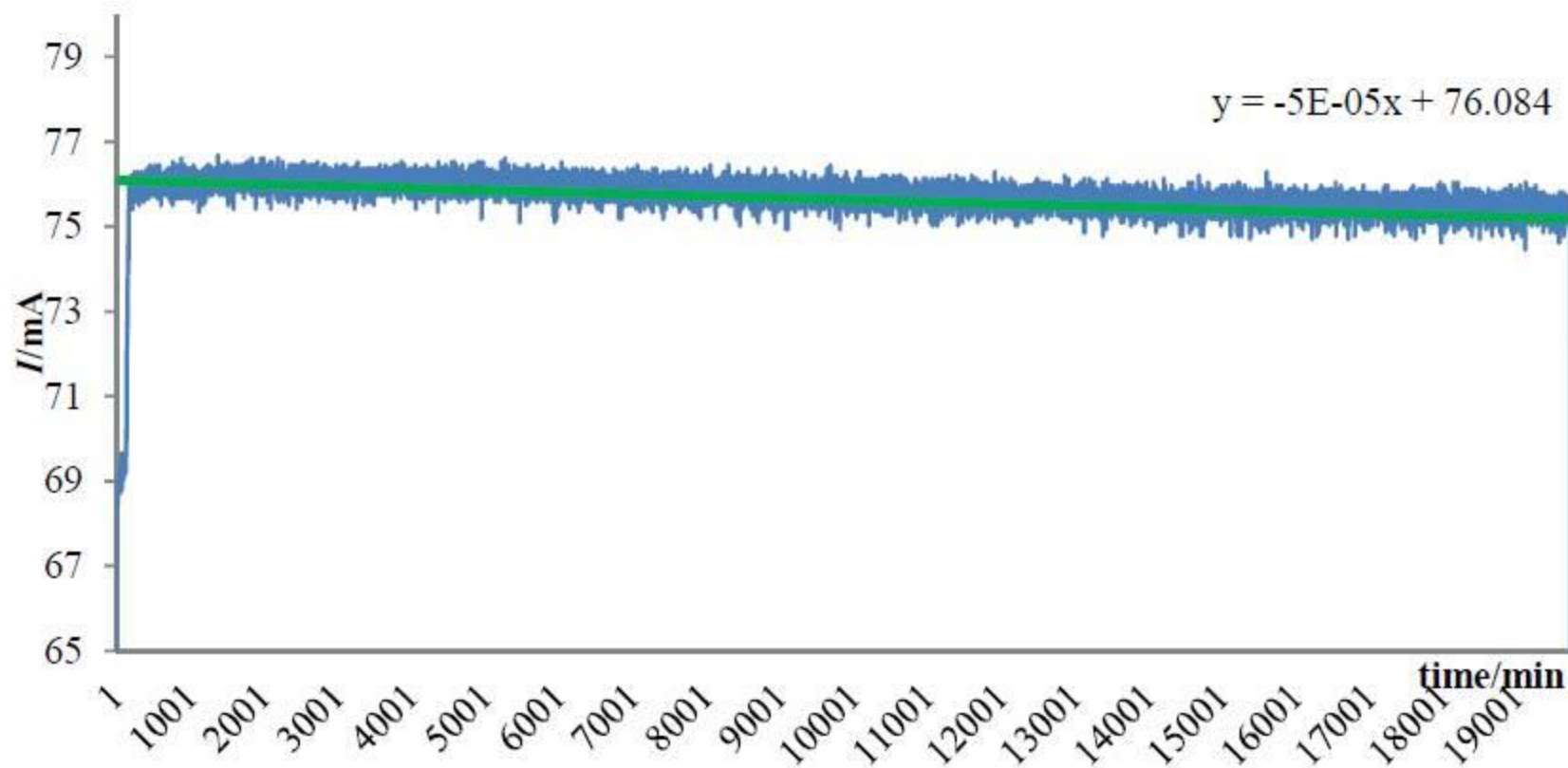


Figure 3 Recorded current intensity I during one 30-houe electrolysis

Figure 4 Close-up on beginning and ending

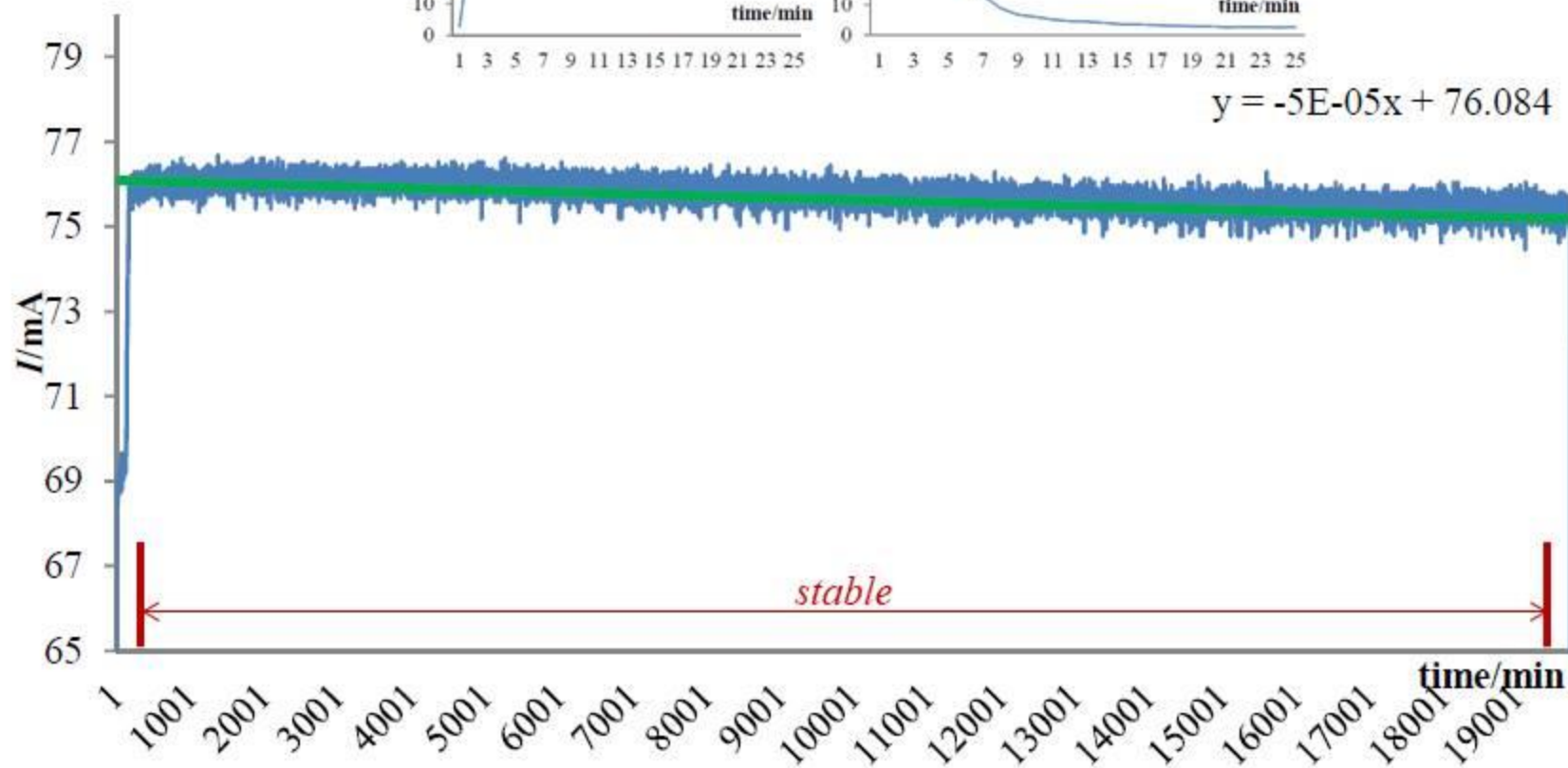
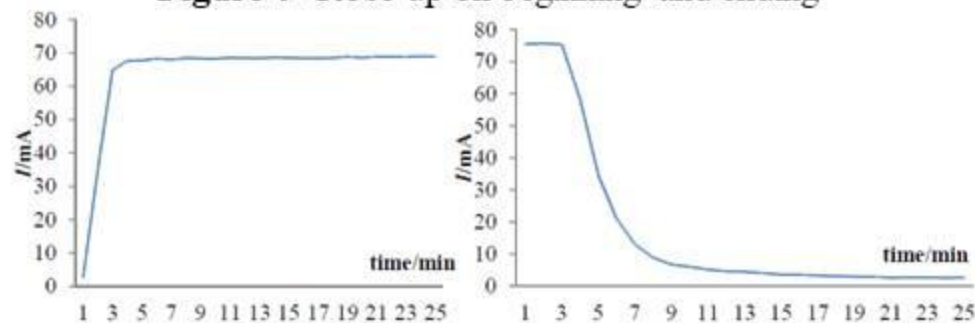


Figure 3 Recorded current intensity I during one 30-houe electrolysis

$N_A(1)$	6.0471×10^{23}	range	$6.0317 \pm 0.0006 \times 10^{23}$
$N_A(2)$	6.0498×10^{23}	real value	6.0221×10^{23}
$N_A(3)$	6.0427×10^{23}	deviation	0.1610557%
$N_A(4)$	6.0206×10^{23}	variance	0.00216689
$N_A(5)$	5.9986×10^{23}		
$\overline{N_A}$	6.0317×10^{23}		

Table 2 Value of N_A with 5 independent electrolysis and the range, deviation and variance

Deviation: $Q \sim 0.01\%$; $m \sim 0.001\%$; $N_A \sim 0.011\%$
 Notes: electrolyte - 45.000g $\text{CuSO}_4 \sim 500.00\text{mL H}_2\text{O}$
 Time: pre-electrolysis > electrolysis

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.....

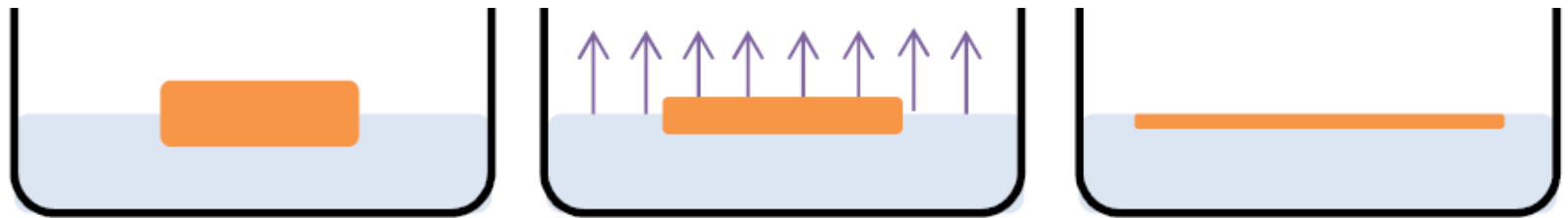


Figure 5 Above illustrates the forming of a monolayer oil film. The vaporization effect of benzene pushes the Stearic acid upwards, stretching through the whole water surface, forming a single layer.

硬脂酸 - 苯 + 爽身粉

$$N_A = \frac{M}{\frac{AV_0m}{SV}} = \frac{MSV}{AV_0m}$$

M : 硬脂酸摩尔质量

S : 单分子油膜面积

V : 苯的体积

m : 硬脂酸质量

A : 单分子油膜面积

V_0 : 单分子油膜体积

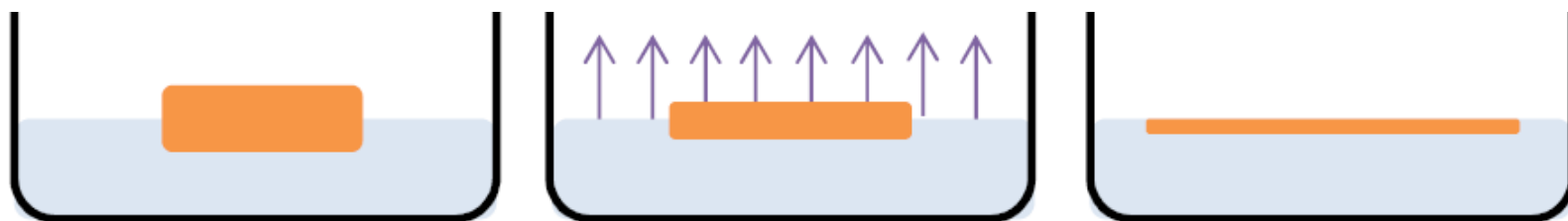


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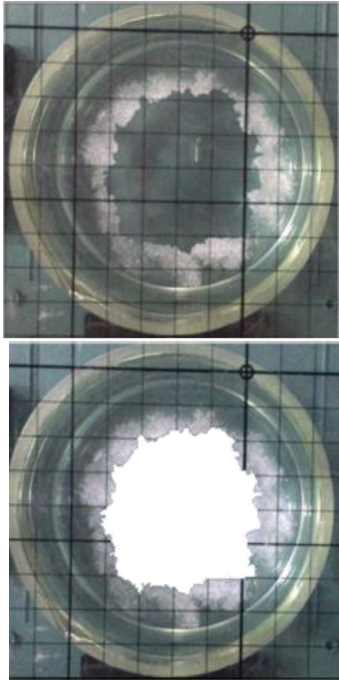


Figure 8 Pixel counting. Indicated with white powder is the basic contour of the oil film formed.

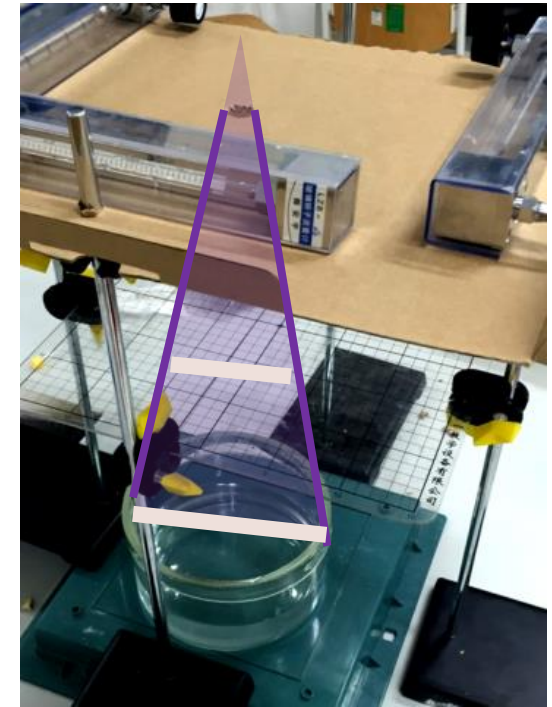
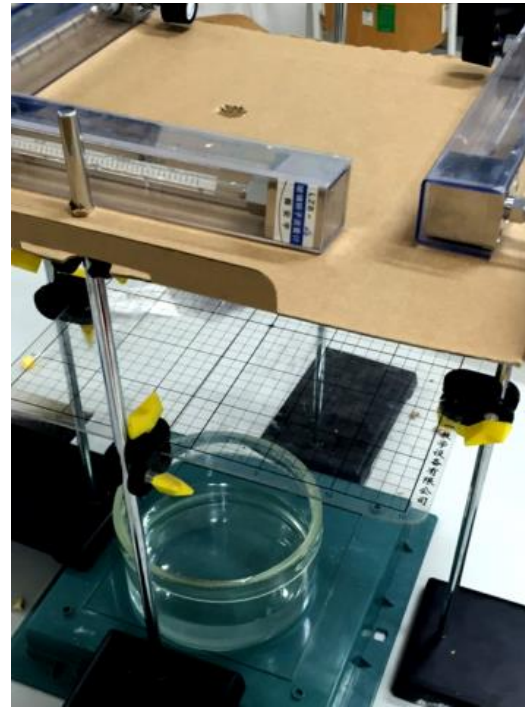


Figure 6 Improved instruments involved in the monolayer oil film

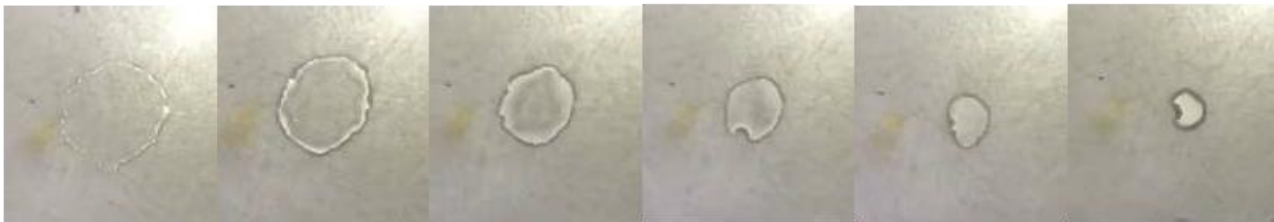


Figure 7 The extension and contraction of a monolayer oil film.



Figure 9 3 typical approximation models adapted investigating the stearic acid molecule

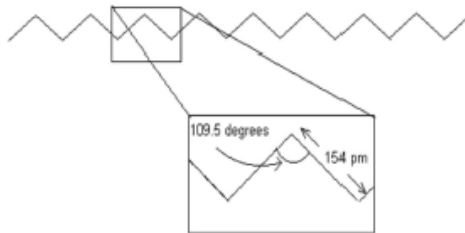


Figure 10 The illustration of a stearic acid molecule

$$M = 284.49 \text{ g mol}^{-1}$$

$$S = 2.789 \times 10^{-4} \text{ m}^2$$

$$V = 1.694 \times 10^{-8} \text{ m}^3$$

Ball:

$$N_A = \frac{6MS^3}{\rho\pi V^3} = 6.747 \times 10^{23}$$

Lying cylinder:

$$N_A = \frac{MS}{lm} = 1.530 \times 10^{23}$$

Standing cylinder:

$$N_A = \frac{4MS}{\pi d^2 \rho V} = 1.355 \times 10^{23}$$

胰岛素- 乙醇

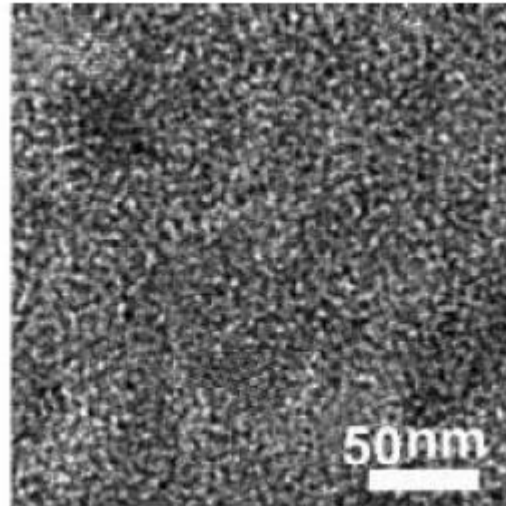


Figure 6 A picture taken during the SEM counting

$$N_A = \frac{nM}{m}, \quad M: \text{胰岛素酸摩尔质量}$$

$$N_A = 6.0237 \times 10^{23}, \text{ deviation} = 0.026\%$$

SEM 计数 > 电解法 > 单分子油膜.

- ✓ 电解法
- ✓ 单分子油膜

谢谢

请各位老师批评指正

Our thank goes to the high school affiliated to RENMIN University of China for providing us with the precious opportunity to conduct independent researches.

Our thank goes to our mentor Zilong Zhao for his continuous support and aid during the researching process.

Our thank goes to the parents who have always been extremely supportive during our researching process.

Our special thanks also go to Mr. Chenbing Wang who has helped us a lot during the development and design of such coulomb meter.

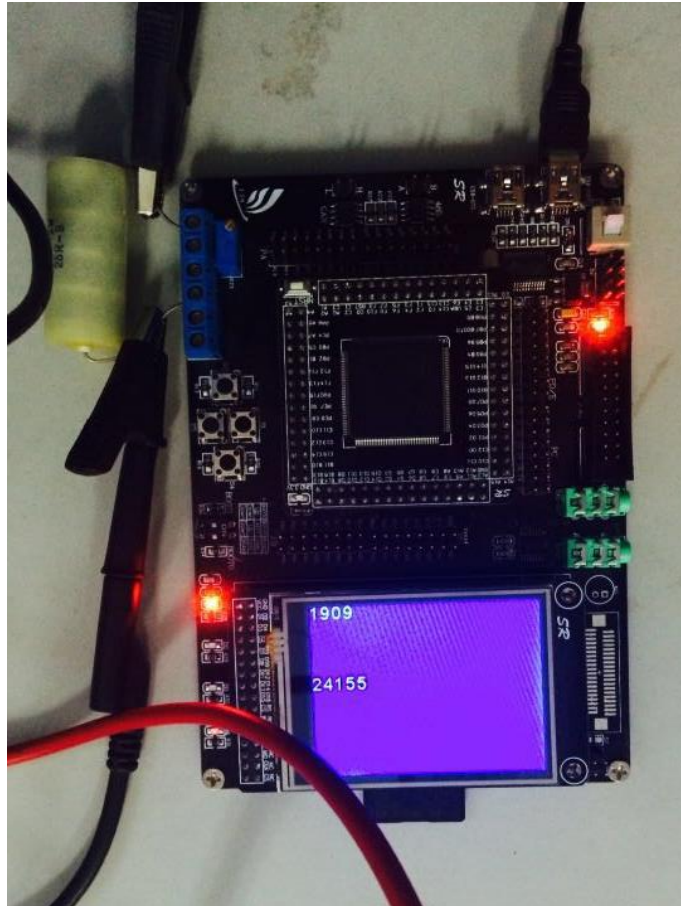
Our thank also goes to the committee of the US Invitational Young Physicist Tournament and every school and student participating in the 2015 competition, for giving us the opportunity to present, discuss, and reflect on our previous work and sparkle new inspiration.

Last but not the least, our thank goes to everyone who has ever lent their helping hand to us during the process.

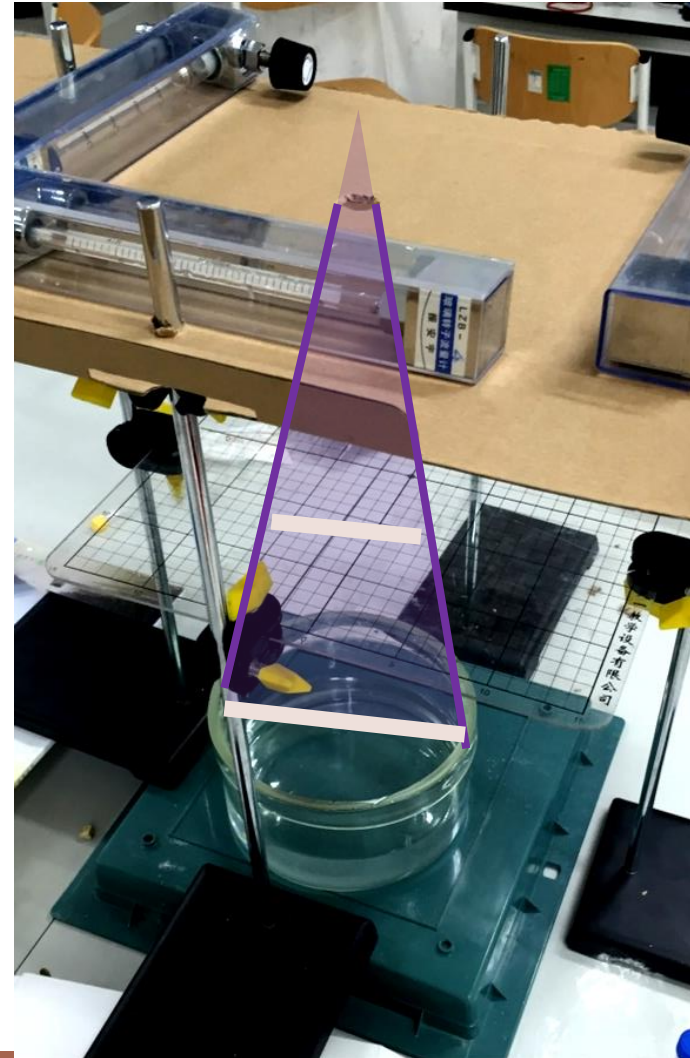
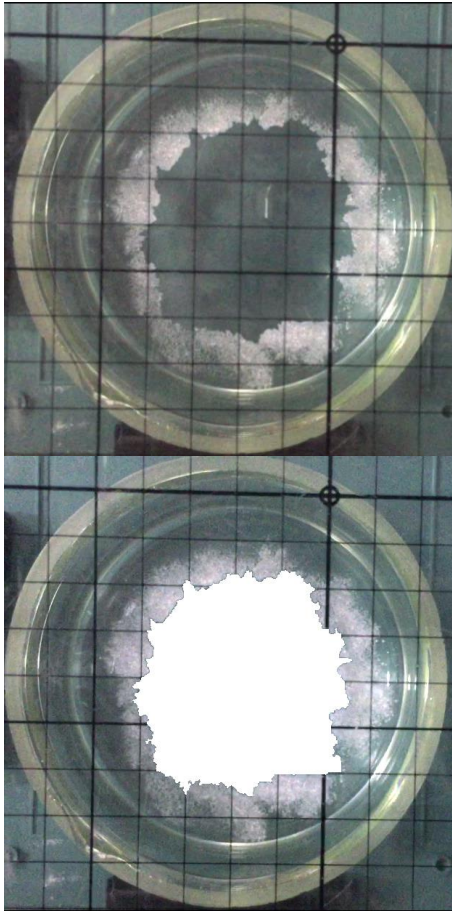
During the researching and investigation process, Wenyan Guan is mostly in charge of the electrolysis experiments, data analysis and paper writing; Zimeng Yang is mostly responsible the monolayer oil film experiment; and Shaoxun Liu is mostly responsible for the SEM counting experiment.

The majority of the experiments are conducted in The High School Affiliated to RENMIN University of China, during Sep.2014-Jan. 2015.

Electrolysis - AD converter

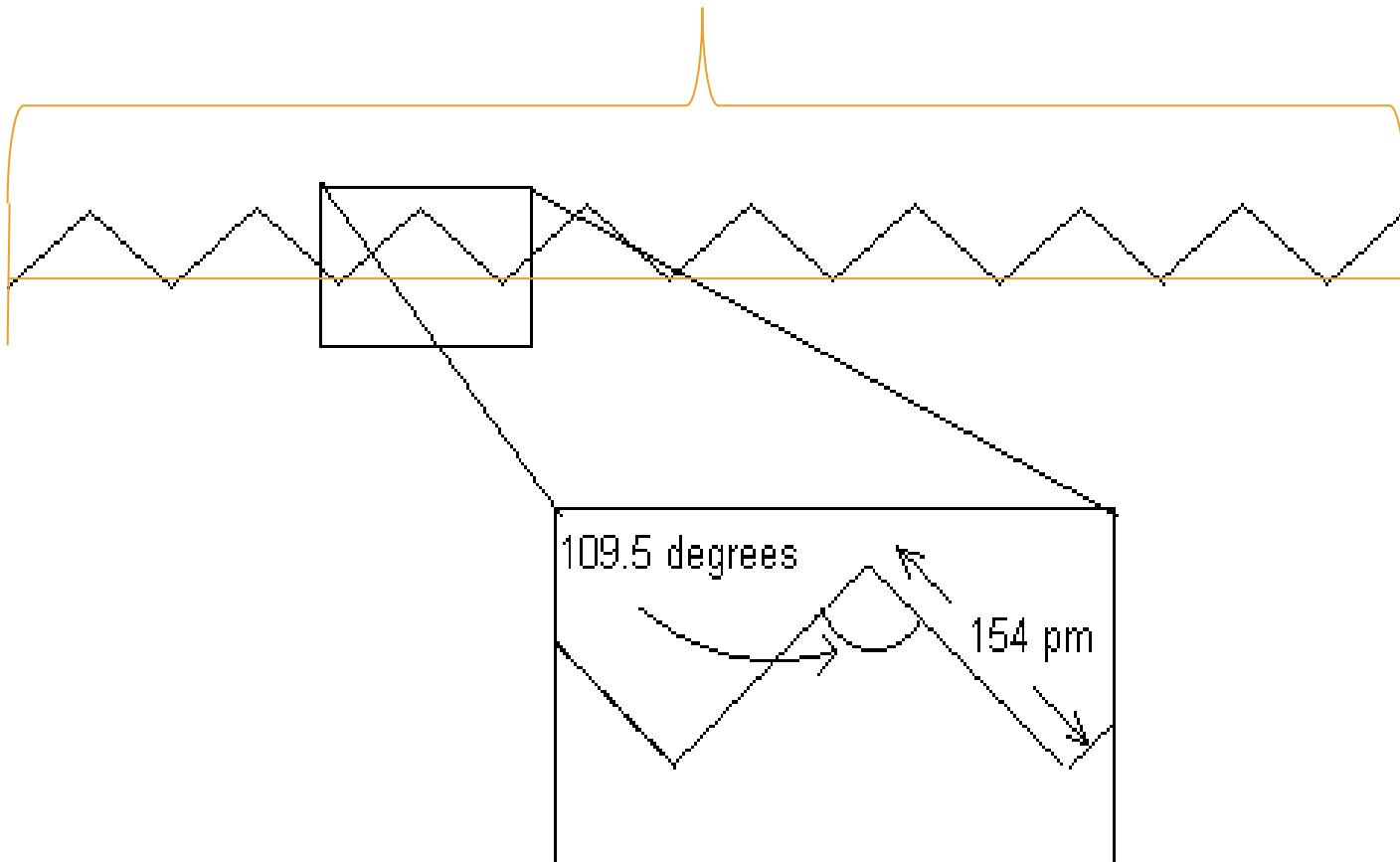


Monolayer - calculating the area



Monolayer - diameter of a single molecule

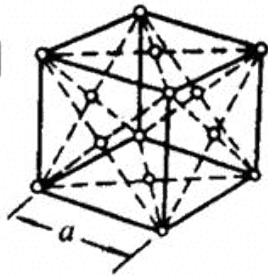
21.37967591 Å



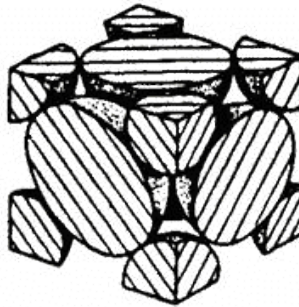
X-ray diffraction



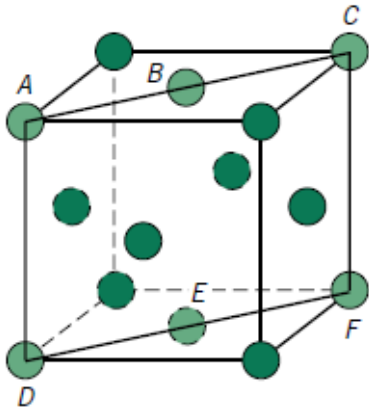
model



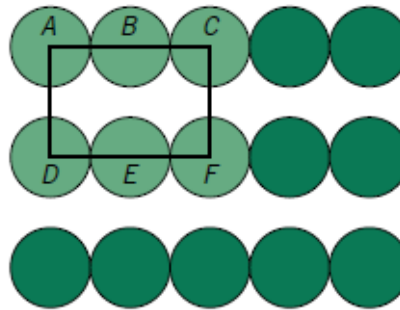
unit cell



structure



(a)



(b)

